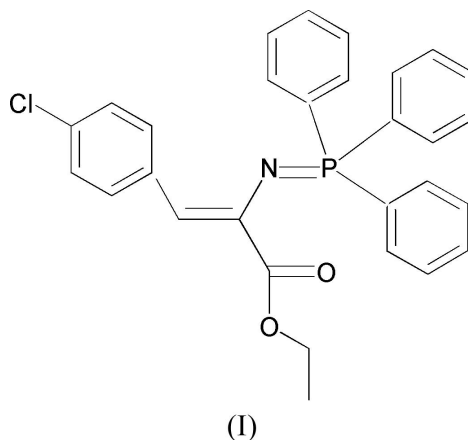


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xiaobhuang@hotmail.com**Key indicators**Single-crystal X-ray study
 $T = 298\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.006\text{ \AA}$
 R factor = 0.078
 wR factor = 0.153
Data-to-parameter ratio = 14.7For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.**(Z)-Ethyl 3-(4-chlorophenyl)-2-[(triphenylphosphoranylidene)amino]prop-2-enoate**The title compound, $\text{C}_{29}\text{H}_{25}\text{ClNO}_2\text{P}$, exists in the *Z* form. The short $\text{C}\cdots\text{O}$ [2.700 (4) Å] and $\text{C}\cdots\text{N}$ [2.963 (4) Å and 2.988 (5) Å] intramolecular contacts may indicate the presence of weak intramolecular hydrogen bonds.

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CommentThe readily available iminophosphoranes have become useful building blocks in organic synthetic strategies directed towards the synthesis of nitrogen-containing heterocycles (Fresneda & Molina, 2004). The title compound, (I), is an intermediate in the preparation of imidazolinones (Ding *et al.* 2003), some of which exhibit fungicidal and herbicidal activities (Yang *et al.*, 2004).The molecule of (I) contains four essentially planar phenyl rings, three of which, C12–C17 (*A*), C18–C23 (*B*) and C24–C29 (*C*), belong to the triphenylphosphine group (Fig. 1). The dihedral angles *A/B*, *A/C* and *B/C* are 56.2 (1), 78.8 (2) and 87.4 (1)°, respectively. Bond lengths and angles in the title compound (Table 1) are similar to those found in (*Z*)-ethyl 3-phenyl-2-[(triphenylphosphoranylidene)amino]prop-2-enoate (Huang *et al.*, 2005).The short $\text{C}\cdots\text{O}$ and $\text{C}\cdots\text{N}$ intramolecular contacts (Table 2) may indicate the presence of weak intramolecular $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{N}$ hydrogen bonds.**Experimental**The title compound was synthesized in 70% yield by the Staudinger reaction of ethyl β -azidoacetate with triphenylphosphine at room temperature (Molina *et al.*, 1993). Single crystals of (I) suitable for X-ray data collection were obtained by slow evaporation of a solution in ethanol and trichloromethane (5:1, *v/v*; m.p. 453–455 K). IR (KBr, cm^{-1}): ν 2950, 1671, 1592, 1410, 1231; ^1H NMR (chloroform-*d*, p.p.m.):

δ 8.12–7.20 (*m*, 19H), 6.67 (*d*, 1H, $J = 7.1$ Hz), 3.86 (*q*, 2H, $J = 7.1$ Hz), 1.00 (*t*, 3H, $J = 7.1$ Hz); ^{31}P NMR (chloroform-*d*, p.p.m): δ 8.60 (*s*).

Crystal data

$\text{C}_{29}\text{H}_{25}\text{ClINO}_2\text{P}$
 $M_r = 485.92$
 Monoclinic, $P2_1/n$
 $a = 10.4682$ (8) Å
 $b = 17.7551$ (13) Å
 $c = 14.1264$ (10) Å
 $\beta = 106.058$ (1)°
 $V = 2523.1$ (3) Å³
 $Z = 4$

$D_x = 1.279$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 1467 reflections
 $\theta = 2.2$ – 24.1 °
 $\mu = 0.24$ mm⁻¹
 $T = 298$ (2) K
 Block, colorless
 $0.23 \times 0.19 \times 0.08$ mm

Data collection

Bruker APEX area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Bruker, 2002)
 $T_{\min} = 0.947$, $T_{\max} = 0.981$
 13 251 measured reflections

4535 independent reflections
 3370 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.057$
 $\theta_{\text{max}} = 25.2$ °
 $h = -12 \rightarrow 12$
 $k = -17 \rightarrow 21$
 $l = -14 \rightarrow 16$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.078$
 $wR(F^2) = 0.153$
 $S = 1.14$
 4535 reflections
 308 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0467P)^2 + 1.2582P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.38$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.33$ e Å⁻³
 Extinction correction: none

Table 1

Selected geometric parameters (Å, °).

C11–C1	1.745 (4)	O1–C10	1.452 (4)
P1–N1	1.576 (3)	O2–C9	1.198 (4)
P1–C29	1.811 (3)	N1–C8	1.369 (4)
P1–C17	1.811 (3)	C7–C8	1.353 (4)
P1–C23	1.812 (3)	C8–C9	1.505 (5)
O1–C9	1.343 (4)		
N1–P1–C29	104.92 (15)	C9–O1–C10	115.9 (3)
N1–P1–C17	115.30 (15)	C8–N1–P1	130.6 (2)
C29–P1–C17	106.53 (15)	C7–C8–N1	123.6 (3)
N1–P1–C23	117.03 (16)	N1–C8–C9	117.6 (3)
C29–P1–C23	101.32 (15)	O2–C9–O1	123.1 (3)
C17–P1–C23	110.04 (15)	O1–C9–C8	113.9 (3)

Table 2

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C28–H28 \cdots N1	0.93	2.59	2.988 (5)	106
C7–H7 \cdots O1	0.93	2.27	2.700 (4)	107
C3–H3 \cdots N1	0.93	2.35	2.963 (4)	123

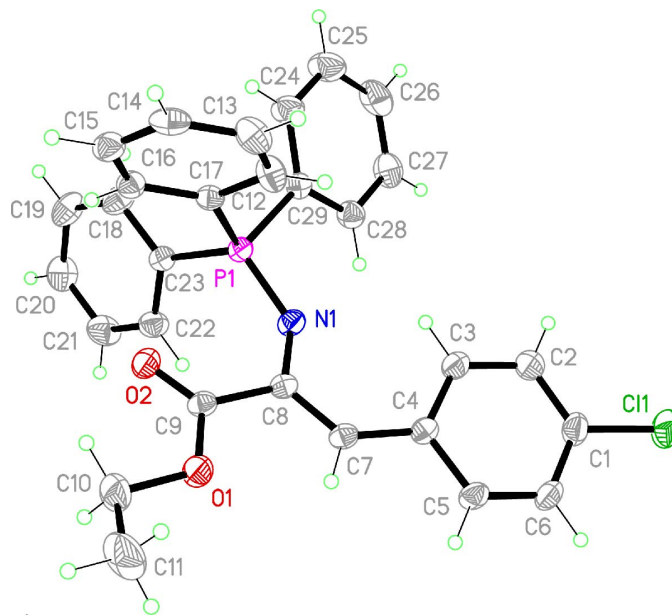


Figure 1

The molecular structure of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

H atoms were positioned geometrically and allowed to ride on their parent atoms at distances of C–H = 0.93 Å (CH) and 0.97 Å (CH₂), with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$, or $\text{Csp}^3\text{–H} = 0.96$ Å, with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$.

Data collection: SMART (Bruker, 2002); cell refinement: SAINT (Bruker, 2002); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 2002); software used to prepare material for publication: SHELXL97.

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References

- Bruker (2002). SADABS (Version 2.03), SAINT (Version 6.02), SMART (Version 5.62) and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
- Ding, M. W., Sun, Y. & Liu, Z. J. (2003). *Synth. Commun.* **33**, 1267–1274.
- Fresneda, P. M. & Molina, P. (2004). *Synlett*, **1**, 1–17.
- Huang, X. B., Liu, M. C., Wu, H. Y., Ding, J. C. & Hu, M. L. (2005). *Acta Cryst. E61*, o280–o281.
- Molina, P., Pastor, A. & Vilaplana, M. J. (1993). *Tetrahedron*, **49**, 7769–7778.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Yang, F. L., Liu, Z. J., Huang, X. B. & Ding, M. W. (2004). *J. Heterocycl. Chem.* **41**, 77–83.